BBA Report

Modulation of Ca²⁺-dependent K ⁺ transport by modifications of the NAD ⁺/NADH ratio in intact human red cells

J. Alvarez, J.M. Camaleño, J. García-Sancho and B. Herreros

Departamento de Fisiología y Bioquímica, Facultad de Medicina, 47005-Valladolid (Spain)

(Received December 2nd, 1985)

Key words: K⁺ transport; Ca²⁺ effect; Redox state; (Human erythrocyte)

The effects of variations of the NAD $^+$ /NADH quotient on the uptake of 86 Rb by human red cells loaded by non-disruptive means with the chelator Benz2 and different amounts of 45 Ca has been examined. The NAD $^+$ /NADH quotient was modified by the addition of pyruvate and/or lactate or xylitol. It was found that the uptake of 86 Rb at a given intracellular Ca^{2+} concentration was faster in the reduced state (lactate or xylitol added). Metabolic changes were associated with variations of the redox state. However, glycolitic intermediates did not significantly modify the apparent affinity for Ca^{2+} of the Ca^{2+} -dependent K^+ channel in one-step inside-out vesicles prepared from the erythrocyte membrane. Taken together, these results suggest that modifications of the cytoplasmic redox potential could modulate the sensitivity to Ca^{2+} of the Ca^{2+} -dependent K^+ channel in the human red cells under physiological conditions. This conclusion is consistent with previous findings in inside-out vesicles of human erythrocytes using artificial electron donors.

It has been shown previously that several redox agents can modify the sensitivity to Ca²⁺ of the Ca2+-dependent K+ channel in the human erythrocyte [1,2]. Using inside-out vesicles we have shown that the reduction of a membrane component with $E'_{o} = 47 \text{ mV}$ (pH 7.5) conferred high affinity for Ca2+ to the K+ channel [3]. Most of these data were obtained using artificial electron donor systems. In order to test the possible physiological relevance of the above observations we have studied here the effects of variations of the NAD⁺/NADH ratio in minimally disturbed intact erythrocytes. These variations were achieved by incubation with lactate or pyruvate or with xylitol, which is able to reduce NAD⁺ in the human erythrocyte [4].

The main technical problem of this study was to achieve and measure stable and uniform intracellular Ca²⁺ concentrations at the submicromolar level, where the threshold for activation of the K⁺ channel lies. This problem was solved by

using cells first loaded by non-disruptive means with the Ca²⁺ chelator Benz2 [5]. The cells were then incubated with ⁴⁵Ca (the total amount always smaller than that of the chelator) in the presence of the divalent cation ionophore A23187. Under these conditions the intracellular Ca2+ concentration can be calculated from the chelator and ⁴⁵Ca contents [5]. Preliminary experiments showed that the intracellular .Ca2+ concentration stabilized within the first 2 min of incubation. The activity of the Ca2+-dependent K+ channel was followed by measuring the uptake of 86Rb added to the incubation medium. The cell contents of 45 Ca and ⁸⁶Rb were measured by differential scintillation counting after several incubation periods terminated by dilution with 10 vols. of ice-cold medium containing 0.1 mM EGTA and 1 mM quinine and inmediate centrifugation through dibutylphthalate oil [6]. Details on the experimental methods are given with the figure legends. The Ca2+-induced uptake of ⁸⁶Rb could be described by single ex-

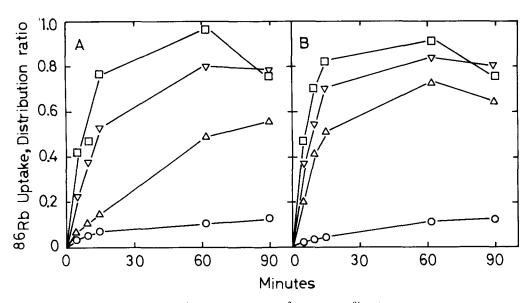


Fig. 1. Effects of variations of NAD⁺/NADH ratio on Ca²⁺-dependent ⁸⁶Rb⁺ uptake by human red cells. Cells were first loaded with Benz2 [5] by incubation at 10% haematocrit and 37°C in standard medium containing 0.1 mM EGTA, 10 mM glucose, 10 mM pyruvate and 0.2 mM Benz2-acetoxymethyltetraester (a generous gift of Dr. V.L. Lew, Physiological Laboratory, Cambridge, U.K.) during 60 min. Standard medium had the following composition (mM): 75 NaCl, 75 KCl, 0.2 MgCl₂, 10 K-Hepes (pH 7.5). Pyruvate was added in order to prevent ATP depletion on loading with the Ca²⁺ chelator [7,8]. After washing, the cells were resuspended at 10% haematocrit in standard medium containing either 10 mM pyruvate (oxidized state, panel A) or 10 mM lactate, 0.7 mM NADH and 0.5 units/ml lactate dehydrogenase (reduced state, panel B). In the last condition NADH and lactate dehydrogenase were added to the medium to secure that any produced pyruvate is quickly reduced to lactate. Extracellular lactate or pyruvate enter the cells and modifies the NAD+/NADH ratio by mediation of the intracellular lactate dehydrogenase [4]. After 60 min at 37°C 1-ml aliquots of the cell suspension were distributed in Eppendorf tubes containing ⁸⁶Rb (about 5·10⁵ cpm/ml) and different ⁴⁵Ca concentrations (specific activity about 10⁶ cpm/μmol). The uptake experiment was started by the addition of the ionophore A23187 (final concentration 10 µM). After different incubation periods, 0.1-ml samples were mixed with 1 ml of ice-cold standard medium containing 0.2 mM EGTA and 1 mM quinine and centrifuged over dibutylphthalate oil [6]. The cell pellets were extracted with 0.4 N perchloric acid and counted for 86Rb and 45Ca radioactivity by differential liquid scintillation counting. The content of Benz2 in the cells was estimated by titration of cell lysates with CaCl₂ under a Ca²⁺-selective electrode. It ranged between 0.4 and 1 mmol/l cells in different experiments. The Ca²⁺ concentration inside the cells was calculated from the ⁴⁵Ca and Benz2 contents assuming for the chelator a dissociation constant of 60 nM [5]. The ⁴⁵Ca content of the cells remained constant for each condition in all the samples taken a different times and it was the same for pyruvate or lactate-treated cells. Symbols represent the following Ca²⁺ concentrations (p*Ca*): < 9 (\bigcirc); 7.40 (\triangle); 6.80 (\triangledown), 3.20 (\square).

ponentials from which the first-order rate constant (k) can be calculated. The amount of ⁸⁶Rb taken up at equilibrium was the same in all the cases, but the rate of uptake increased with increasing Ca^{2+} concentrations.

Fig. 1 shows the uptake of ⁸⁶Rb by Benz2-loaded cells first incubated with either pyruvate (A) or lactate (B) at different Ca²⁺ concentrations. The preincubation with pyruvate or lactate did not significantly modify the intracellular Ca²⁺ concentration obtained. At maximal Ca²⁺ or without Ca²⁺ (upper and lower curves) the rate of uptake was the same in both conditions, but at

intermediate Ca²⁺ concentrations the uptake was faster in the cells incubated with lactate. In another set of experiments the decrease of the NAD⁺/NADH ratio was accomplished using xylitol instead of lactate. Xylitol is able to reduce NAD⁺ through xylitol dehydrogenase and NADH remains reduced if no other substrates are added to the incubation medium [4]. The results of one of such experiments are shown in Fig. 2. The uptake was faster in xylitol-treated cells at all the Ca²⁺ concentrations tested.

In order to obtain a continuous range of redox potentials, aliquots of a cell suspension were first

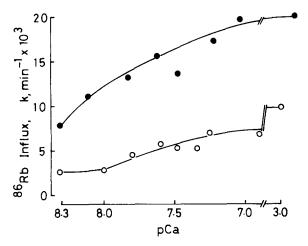
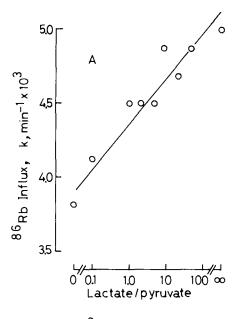


Fig. 2. Effects of treatment with xylitol on the rate of ⁸⁶Rb⁺ uptake at different cell Ca²⁺ concentrations. Cells were loaded with Benz2 as described in the legend to Fig. 1 and incubated either with 10 mM pyruvate (open symbols) or 10 mM xylitol (closed symbols) during 60 min at 37°C before the uptake experiments were started. Other details as in Fig. 1.

incubated with lactate and pyruvate at different ratios. Then a fixed amount of Ca²⁺ giving submaximal activation was added to all the cell suspensions and the uptake of ⁸⁶Rb was measured. Fig. 3A shows the changes of the rate of uptake as a function of the ratio lactate/pyruvate. The rate of uptake tended to increase with the increase of this ratio, the measured calcium content of the cells remaining unmodified (not shown). The cell ATP contents of these cells by the end of the experiment were also measured, and were found to decrease as the ratio lactate/pyruvate increased (Fig. 3B).

All the above observations support the idea that changes of the NAD⁺/NADH ratio that could take place under physiological circumstances can modigy the sensitivity to Ca^{2^+} of the Ca^{2^+} -dependent K⁺ channel. These modifications should not be expected, in principle, from the $E'_{\rm o}$ of 47 mV (pH 7.5) previously determined for the membrane-associated component in insideout vesicles [4], since $E'_{\rm o}$ of the lactate/pyruvate pair is much more negative (-210 mV at pH 7.5). These results, however, are not necessarily conflicting if additional electron transfer steps were involved in the intact cell or the membrane carrier were somehow modified during the preparation of

inside-out vesicles. It should be noted, however, that the effects of changes of the NAD⁺/NADH ratio on Ca²⁺-dependent ⁸⁶Rb transport reported



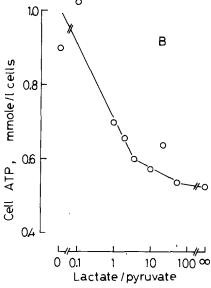


Fig. 3. Effects of variations of the lactate/pyruvate ratio on the rate of uptake of $^{86}\mathrm{Rb^+}$ (A) or the ATP content (B) of human red cells at a fixed intracellular $\mathrm{Ca^{2+}}$ concentration. Cells loaded with Benz2 were incubated during 60 min at $37^{\circ}\mathrm{C}$ at different lactate/pyruvate ratios as indicated in the figure. The same amount of Ca was added for the uptake experiment in all the cases, and the measured $\mathrm{Ca^{2+}}$ concentration in the cells did not show any trend of variation in the different conditions. Mean \pm S.D. of all the nine pCa values was 7.65 ± 0.04 .

here are much weaker than those reported previously using artifitial electron donors [1–3]. The results reported here are consistent with the older observation that pyruvate prevents the activation of K^+ transport in fluoride-poisoned cells [9].

Unfortunately, there was no way to avoid other metabolic changes associated to the variation of the NAD+/NADH quotient. For example, a decrease of the cell ATP level associated to the decrease of the pyruvate/lactate ratio was shown in Fig. 3B. Such changes could be responsible of the observed modifications of the Ca²⁺- sensitivity. In order to test this point, the effect of several glycolitic intermediates on the Ca²⁺- sensitivity of ⁸⁶Rb uptake was tested in one-step inside-out vesicles [10]. The methodology previously described [3,11] was used, except that Mg²⁺ was routinely included in the incubation medium at 5 mM. The following metabolic intermediates, tested at concentrations of 200 to 400 µM, were without effect on Ca²⁺-dependent ⁸⁶Rb uptake by insideout vesicles: glucose 1-phosphate, glucose 6-phosphate, fructose 6-phosphate, fructose 1,6-bisphosphate, 3-phosphoglyceric acid, 2-phosphoglyceric acid, phosphoenol pyruvic acid, ATP, ADP, AMP, cyclic AMP, GTP and UTP. 2,3-Diphosphoglycerate at 4 mM decreased the rate of uptake of ⁸⁶Rb at submaximal Ca²⁺ concentrations to about 1/3 of the control value without modifying the fraction of activated vesicles reached at the steady state. This inhibitory effect of diphosphoglycerate was not observed when the Mg²⁺ concentration of the medium was lowered to 0.5 mM.

The changes of Ca²⁺-dependent K⁺ transport observed on treatment of the intact cells with pyruvate, lactate or xylitol could not be, however,

attributed to modifications of the cell levels of 2,3-diphosphoglycerate since: (i) changes of 2,3-diphosphoglycerate levels should be very small under our experimental conditions [4], and (ii) at the cellular Mg²⁺ level of about 0.5 mM, which should apply to our experiments [12], little or no effect of 2,3-diphosphoglycerate was observed in inside-out vesicles. These results reinforce our conclusion that the Ca²⁺-dependent K⁺ channel of the human erythrocyte can be subjected to redox regulation under physiological conditions.

Support from the spanish Fondo Nacional para el Desarrollo de la Investigación Científica (Proyecto No. 2873/83) is gratefully acknowledged.

References

- 1 Garcia-Sancho, J., Sanchez, A. and Herreros, B. (1979) Biochim. Biophys. Acta 556, 118-130
- 2 Garcia-Sancho, J. and Herreros, B. (1983) Cell Calcium 4, 493-497
- 3 Alvarez, J., Garcia-Sancho, J. and Herreros, B. (1984) Biochim. Biophys. Acta 771, 23-27
- 4 Momsen, G. (1981) Arch. Biochem. Biophys. 210, 160-166
- 5 Lew, V.L., Tsien, R.Y., Miner, C. and Bookchin, R.M. (1982) Nature 298, 478-481
- 6 Ferreira, H.G. and Lew, V.L. (1976) Nature, 259, 47-49
- 7 Tiffert, T., Garcia-Sancho, J. and Lew, V.L. (1983) Biochim. Biophys. Acta 773, 143-156
- 8 Garcia-Sancho, J. (1985) Biochim. Biophys. Acta 813, 148–150
- 9 Lepke, S. and Passow, H. (1968) J. Gen. Physiol. 51, 365s-372s
- 10 Lew, V.L., Muallem, S. and Seymour, C.A. (1982) Nature 296, 742-744
- 11 Garcia-Sancho, J., Sanchez, A. and Herreros, B. (1982) Nature 296, 744-746
- 12 Flatman, P.W. and Lew, V.L. (1980) J. Physiol. (London) 305, 13-30